

Tabelle 1. Flüchtige Inhaltsstoffe aus *Adoxa moschatellina*

Verbindung	Flächenprozent*
<i>trans</i> -2-Hexenal (1)	42,0
<i>cis</i> -3-Hexenol (2)	21,0
Benzylalkohol (3)	13,5
<i>n</i> -Hexanol (4)	12,5
2-Hexenol (5)	10,0

\* Die Summe der identifizierten Riechstoffe (ca 20% des Extrakts) wurde als 100 Flächen-% gesetzt. Weiterhin wurden noch gefunden: *n*-Alkane von C-20 bis C-27 sowie Eicosan- und Docosansäuremethylester.

mit exaltierenden Geruchseigenschaften konnten in *A. moschatellina* nicht nachgewiesen werden.

#### EXPERIMENTELLES

**Gaschromatographie.** 50 m Glaskapillare WG 11 (FFAP-Phase). Trägergas: He; Vordruck: 2, 3 bar; Programmierung: 60–220°, Rate 4°/min.

**Massenspektrometrie.** Hewlett-Packard 5992A, GC-MS-

System, 70 eV.

**Gewinnung des Extraktes.** Die frisch gesammelten Blütenstände (10 g) wurden unzerkleinert in einer Soxhlet-Apparatur unter einer Argon-Atmosphäre 8 Stdn. mit 1,1,2-Trichlor-1,2,2-trifluorethan (Kp 47,6°) extrahiert. Nach Abdampfen des Lösungsmittels über eine 15 cm Vigreux-Kolonnen verblieben ca 70 mg Rückstand als gelbes Öl.

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## ANTIMICROBIAL ACTIVITY OF THE HELIANGOLIDE CHROMOLAENIDE AND RELATED SESQUITERPENE LACTONES

JOSÉ CALZADA\*†, JOSÉ F. CICCIO\*† and GUILLERMO ECHANDI‡

\*Centro de Investigación de Productos Naturales and †Escuela de Química, Universidad de Costa Rica San José, Costa Rica; ‡Facultad de Medicina Veterinaria, Universidad Nacional Heredia, Costa Rica

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**Key Word Index**—Chromolaenide derivatives; sesquiterpene lactones; heliangolides; antimicrobial activity; proof of stereochemistry.

Some sesquiterpene lactones are known to exhibit antimicrobial activity [1, 2]. More recently, it has been noted that the pseudoguinolide carpesiolin inhibits the growth of *Xanthomonas oryzae* [3]. Here, we report on the antimicrobial activity of the heliangolide chromolaenide **1**, isolated from *Chromolaena glaberrima* (DC.) King et Robinson [4] and its derivatives, against *Staphylococcus aureus*.

The results are summarized in Table 1. It is apparent that there is a clear dependence between antimicrobial activity and the presence of the  $\alpha$ -methylene- $\gamma$ -lactone group. As already established [5–8], the growth inhibition of microorganisms is caused by the alkylation of their nucleophilic centers (R-SH) with the unsaturated lactone moiety. This observation is

congruent with our results, because the only inactive compound was **3**, which is the one with no unsaturated lactone.

Chromolaenide **1** and the corresponding acetate **2** were the more active. This is reasonable because compounds with an unsaturated ester group close to the unsaturated lactone show enhanced activity due to anchimeric assistance [9]. The acetate of chromolaenide **2** was more potent than chromolaenide **1** itself. Also compound **5** showed more inhibition than **4**. This result can be explained by the fact that acetates are more lipophilic than the related alcohols. This undoubtedly indicates a problem of transport and not a problem of reactivity. Similar observations have been described before in the literature for cytotoxic

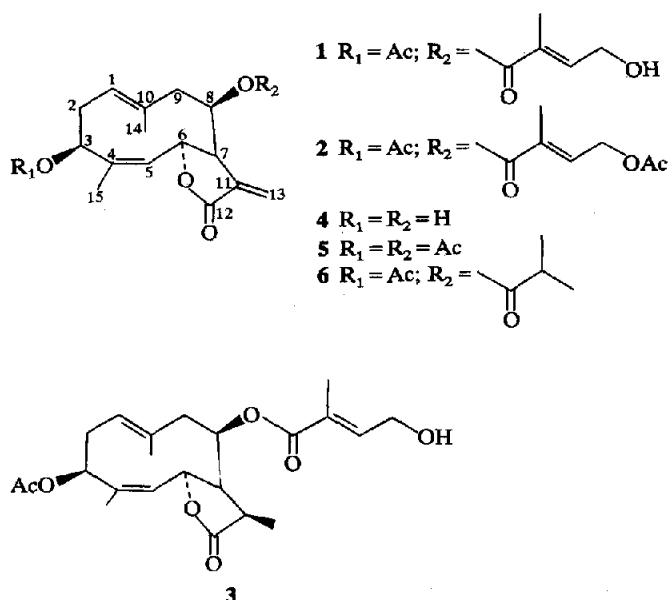


Table 1. Antimicrobial activity of chromolaenide and related compounds

Compound	MIC* ( $\mu\text{g/ml}$ )
1	200
2	100
3	> 1000
4	1000
5	600

\*Minimum inhibitory concentration against *Staphylococcus aureus*. Compounds 1 and 2 were inactive against *Escherichia coli*, *Klebsiella pneumoniae* and *Candida* sp. at the 1000  $\mu\text{g/ml}$  level.

compounds [9, 10].

Compounds 1 and 2 were inactive against *Escherichia coli*, *Klebsiella pneumoniae* (Gram-negative) and *Candida* sp. (yeast), so that it appears that these two lactones are only effective against Gram-positive bacteria.

#### EXPERIMENTAL

**Isolation of chromolaenide.** The heliangolide 1 was extracted from the aerial part of the plant *Chromolaena glaberrima* (DC.) King et Robinson collected in June 1977, in Alajuela, Costa Rica. The dried material was extracted and worked according to the method described by Herz [11]. Chromolaenide 1 was eluted with  $\text{C}_6\text{H}_6$ -EtOAc (4:1) from a Si gel column and crystallized from petrol- $\text{CH}_2\text{Cl}_2$  as colourless prisms, mp 144–146° (lit. [4] mp 138°). The  $^1\text{H}$  NMR, IR and MS were in complete agreement with those reported [4].

**Derivatives of chromolaenide.** Acetylation of lactone 1 produced the diacetate 2, mp 127–128° (lit. [4] mp 127°). The reduction of 1 with  $\text{NaBH}_4$  in MeOH at 0° afforded 3 as a colourless oil. These two compounds showed spectroscopic data consistent with those in the literature [4]. The diol 4 was prepared by saponification of 1 according to the procedure of Mabry [12]. Acetylation of the diol gave rise to the diacetate 5. The  $^1\text{H}$  NMR of 5 is identical to the one obtained from the diacetate prepared from peucephyllin 6 by a similar series of reactions [13–14]. This represents unequivocal proof of the stereochemistry of chromolaenide 1. The structure and relative stereochemistry of peucephyllin were fully established by single crystal X-ray analysis [15].

**Microorganisms.** The organisms utilized in this study included (a) *Staphylococcus aureus* ATCC 6538P (Gram-positive), (b) *Escherichia coli* isolated from a case of colibacillosis in chickens (Gram-negative), (c) *Klebsiella pneumoniae* ATCC 10031 (Gram-negative), (d) *Candida* sp. (yeast), a donation from Facultad de Microbiología, Universidad de Costa Rica.

**Antimicrobial testing.** The determination of the minimum inhibitory concn was carried out according to the method

developed by Mitscher [15].

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